

Solvent Deasphalting



- In this extraction process, which uses propane (or hexane) as a solvent, heavy oil fractions are separated to produce heavy lubricating oil, catalytic cracking feedstock, and asphalt.
- Feedstock and liquid propane are pumped to an extraction tower at precisely controlled mixtures, temperatures (150°-250° F), and pressures of 350-600 psi.
- Separation occurs in a rotating disc contactor, based on differences in solubility.
- The products are then evaporated and steam stripped to recover the propane, which is recycled.
- Deasphalting also removes some sulfur and nitrogen compounds, metals, carbon residues, and paraffins from the feedstock.

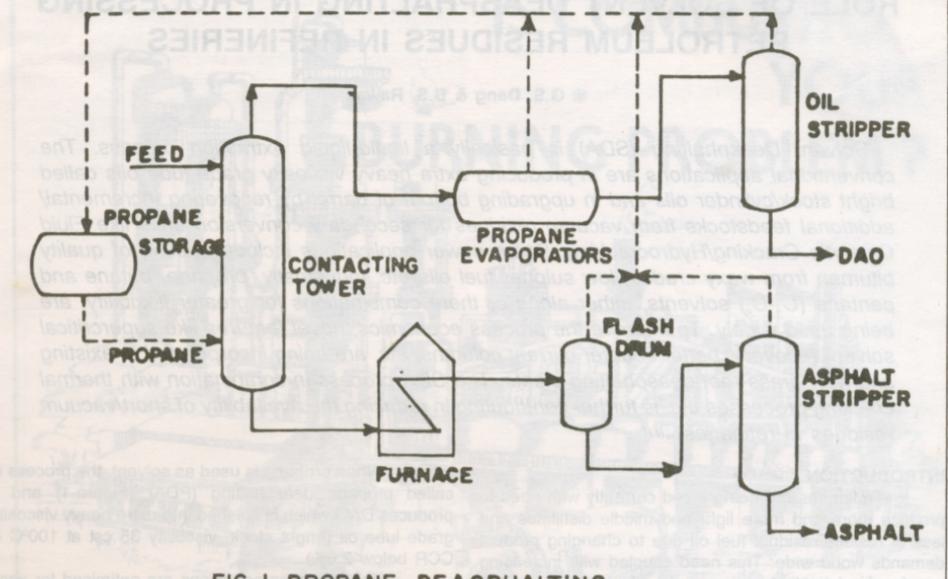


FIG. I-PROPANE DEASPHALTING



Solvent Deasphalting



Feedstock	From	Process	Typical products To
Residual, reduced crude	Atmospheric tower & Vacuum tower	Treatment	Heavy lube oil Treating or lube blending Asphalt Storage of shipping Deasphalted oil Hydrotreat & catalytic cracker Propane Recycle

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Solvent Dewaxing



- Solvent dewaxing is used to remove wax from either distillate or residual basestocks at any stage in the refining process.
- There are several processes in use for solvent dewaxing, but all have the same general steps, which are:
 - (1) mixing the feedstock with a solvent,
 - (2) precipitating the wax from the mixture by chilling, and
 - (3) recovering the solvent from the wax and dewaxed oil for recycling by distillation and steam stripping.
- Usually two solvents are used: toluene, which dissolves the oil and maintains fluidity at low temperatures, and methyl ethyl ketone (MEK), which dissolves little wax at low temperatures and acts as a wax precipitating agent.



Solvent Dewaxing



- Other solvents that are sometimes used include benzene, methyl isobutyl ketone, propane, petroleum naphtha, ethylene dichloride, methylene chloride, and sulfur dioxide.
- In addition, there is a catalytic process used as an alternate to solvent dewaxing.

Feedstock	From	Process	Typical products To
Lube basestock	Vacuum tower	Treating	Dewaxed lubes Hydrotreating Wax Hydrotreating Spent agents

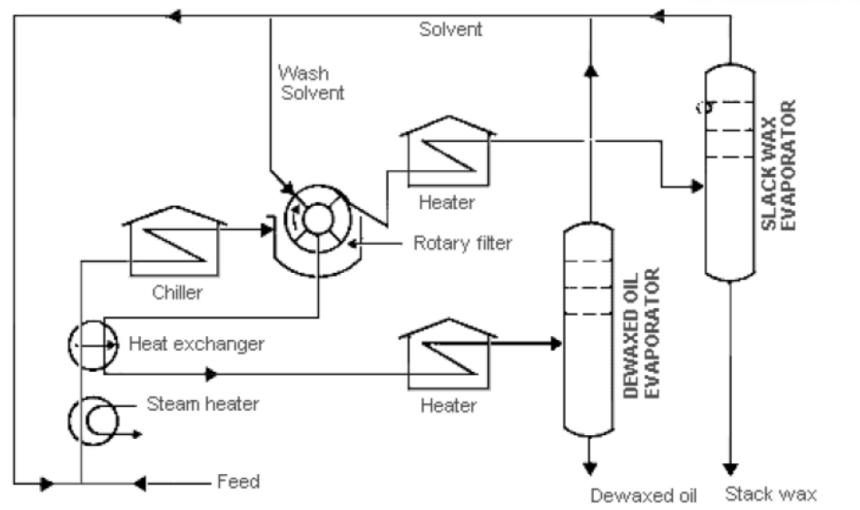
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Solvent Dewaxing









- The purpose of solvent extraction is to prevent corrosion, protect catalyst in subsequent processes, and improve finished products by removing unsaturated, aromatic hydrocarbons from lubricant and grease stocks.
- The solvent extraction process separates aromatics, naphthenes, and impurities from the product stream by dissolving or precipitation.
- The feedstock is first dried and then treated using a continuous countercurrent solvent treatment operation.
- In one type of process, the feedstock is washed with a liquid in which the substances to be removed are more soluble than in the desired resultant product.
- In another process, selected solvents are added to cause impurities to precipitate out of the product.





- In the adsorption process, highly porous solid materials collect liquid molecules on their surfaces.
- The solvent is separated from the product stream by heating, evaporation, or fractionation, and residual trace amounts are subsequently removed from the raffinate by steam stripping or vacuum flashing.
- Electric precipitation may be used for separation of inorganic compounds.
- The solvent is then regenerated to be used again in the process.
- The most widely used extraction solvents are phenol, furfural, and cresylic acid.





- Other solvents less frequently used are liquid sulfur dioxide, nitrobenzene, and 2,2'-dichloroethyl ether.
- The selection of specific processes and chemical agents depends on the nature of the feedstock being treated, the contaminants present, and the finished product requirements.





Feedstock	From	Process	Typical products To	
Naphthas,	Atm. tower	Treating/	High octane gasoline Storage	
distillates, kerosene		blending	Refined fuels Treating and blending Spent agents Treatment and blending	





